# organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 292 KMean  $\sigma(\text{C-C}) = 0.003 \text{ Å}$  R factor = 0.056 wR factor = 0.159 Data-to-parameter ratio = 16.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 8-Butylamino-1,2,5-trimethyl-7-*p*-tolyl-3-thia-4,7,9-triazacyclopenta[*a*]naphthalen-6(7*H*)-one

In the title compound,  $C_{23}H_{26}N_4OS$ , the fused tricyclic system is essentially planar. The crystal stacking is mainly due to  $\pi$ - $\pi$ interactions and intermolecular N-H···O hydrogen-bond interactions. Received 30 August 2005 Accepted 30 September 2005 Online 12 October 2005

### Comment

Among many known heterocyclic compounds, derivatives containing the pyridine system have received much attention since they have been demonstrated to possess significant biological activity (Augusto *et al.*, 1995). The title compound, (I), belongs to this family of heterocyclic compounds and we present its crystal structure here.



In (I) (Fig. 1), the C–S bond lengths [1.733 (2) and 1.741 (2) Å] are greater than those observed in free thiophene [1.714 (3) Å; Bonham & Momany, 1963] and thieno[2,3-*c*]pyridine [1.728 (1) and 1.731 (1) Å; Nerenz *et al.*, 1997]. Selected bond lengths and angles are listed in Table 1. The C3–S1–C6 angle of 91.43 (10)° in (I) is slightly less than that observed in free thiophene [92.2 (2)°]. As expected for a non-



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## Figure 1

View of (I), showing the labelling scheme for the non-H atoms and 50% probability displacement ellipsoids.



Figure 2

The crystal structure of (I), showing the formation of N-H···O hydrogen bonds (dashed lines).

protonated ring system, the C12-N2-C10 angle of  $118.22 (18)^{\circ}$  is smaller than  $120^{\circ}$  (Ghosh & Simonsen, 1993). The torsion angles S1-C6-C5-C10 and N2-C10-C9-C7 are -178.57 (15) and -176.09 (19)°, respectively, showing the essential planarity of the fused tricyclic system. There are some weak intermolecular N-H···O hydrogen-bonding interactions (Table 2).

The short intermolecular distances between the centroids of the thiophene (Cg1) and pyridine (Cg2) rings  $[Cg1 \cdots Cg2^{i} =$ 3.691 (1) Å and  $Cg2 \cdots Cg2^{i} = 3.843$  (1) Å; symmetry code: (i)  $\frac{1}{2} - x$ ,  $\frac{3}{2} - y$ , -z indicate the existence of  $\pi - \pi$  stacking interactions (Janiak, 2000), which stabilize the crystal packing (Fig. 2) together with the hydrogen-bonding interactions.

## **Experimental**

2,3,6-Trimethyl-4-amino-5-ethoxycarbonylthieno[2,3-b]pyridine, (II), was prepared according to a literature procedure in 63% yield (Augusto et al., 1995). The iminophosphorane of (II) was synthesized according to a literature report in 93% yield (Wamhoff et al., 1993). To a solution of the iminophosphorane of (II) (1 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (20 ml) containing a catalytic amount of K<sub>2</sub>CO<sub>3</sub> (0.05 mmol) was added 4-methylphenyl isocyanate (1.1 mmol) under an N<sub>2</sub> atmosphere at room temperature (Ding et al., 1999). After the reaction mixture was allowed to stand for 5-12 h, the solvent was removed under reduced pressure and Et<sub>2</sub>O and petroleum ether were added to precipitate the side product triphenylphosphine oxide which was then removed by filtration. Subsequent removal of the solvent gave the corresponding carbodiimide, which was used directly without further purification. To a solution of the carbodiimide prepared above in ethanol (15 ml) was added n-butylamine (1.1 mmol) and a catalytic amount of sodium ethoxide in ethanol (Wang et al., 2004). After the mixture had been stirred for 14 h at



Part of the crystal structure of (I), showing the  $\pi$ - $\pi$  stacking interactions.

303 K, the solution was concentrated and the residue was recrystallized from CH<sub>3</sub>CN to give colorless prisms of the title compound, (I), after one week.

Crystal data

C23H26N4OS	$D_x = 1.295 \text{ Mg m}^{-3}$
$M_r = 406.54$	Mo $K\alpha$ radiation
Monoclinic, $C2/c$	Cell parameters from 2860
a = 21.5659 (19)  Å	reflections
b = 9.0711 (8) Å	$\theta = 2.3-25.9^{\circ}$
c = 22.646 (2) Å	$\mu = 0.18 \text{ mm}^{-1}$
$\beta = 109.746 \ (2)^{\circ}$	T = 292 (2) K
V = 4169.7 (6) Å <sup>3</sup>	Prism, colorless
Z = 8	$0.30 \times 0.20 \times 0.10 \text{ mm}$

### Data collection

Bruker SMART CCD area-detector
diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1997)
$T_{\min} = 0.949, T_{\max} = 0.965$
11954 measured reflections
$D = t^{\alpha} \cdots \cdots$

## Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0773P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.056$	+ 3.0234P]
$wR(F^2) = 0.159$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.05	$(\Delta/\sigma)_{\rm max} = 0.003$
4535 reflections	$\Delta \rho_{\rm max} = 0.55 \ {\rm e} \ {\rm \AA}^{-3}$
271 parameters	$\Delta \rho_{\rm min} = -0.29 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	

## Table 1

Sel	lected	geometric	parameters	(A,	°)	).
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C3-S1	1.741 (2)	C6-S1	1.733 (2)
C12-N2-C10	118.22 (18)	C6-S1-C3	91.43 (10)
C10-C5-C6-S1	-178.57 (15)	C7-C9-C10-N2	-176.08 (19)

## Table 2

#### Hydrogen-bond geometry (Å, °).

2.22 (2)	2.948 (2)	139 (2)
2.40	2.814 (3)	106
	2.22 (2) 2.40	2.22 (2)2.948 (2)2.402.814 (3)

All C-bound H atoms were positioned geometrically, with C-H distances of 0.96, 0.93 and 0.97 Å, respectively, for methyl, methylene

4535 independent reflections

 $R_{\rm int} = 0.031$  $\theta_{\rm max} = 27.0^{\circ}$ 

 $h = -27 \rightarrow 25$ 

 $k = -11 \rightarrow 11$ 

 $l = -28 \rightarrow 26$ 

3493 reflections with  $I > 2\sigma(I)$ 

and aromatic H atoms. The H atoms were allowed to ride on their parent atoms, with  $U_{iso}(H) = 1.2U_{eq}(C)$  or  $1.5U_{eq}(methyl C)$ . The N-bound atom H4 was located in a difference Fourier map and refined with the N-H distance restrained to 0.89 (1) Å, with  $U_{iso}(H) = 0.05 \text{ Å}^2$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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